search for a crystal of La₂Sb (the original crystal had decomposed) eventually yielded a workable crystal. Fullmatrix least-squares refinement of 455 reflections [$I > 2\sigma(I)$, $2\theta_{max} = 90^{\circ}$, monochromatized Mo $K\alpha$ radiation, diffractometer data with anisotropic temperature factors and allowance for anomalous dispersion] gave R = 0.086.* (This relatively high value of R reflects the poor quality of the crystal, the best found in an extensive search.) A difference Fourier synthesis showed no residual electron density at the 2(a) or 2(b) sites although there were small ripples close to atom sites. Full details of the structure are given by Stassen, Sato & Calvert (1970).

* Full atomic parameters and a list of structure factors have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34791 (8 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. We conclude that La₂Sb contains no interstitial atom; moreover, it is known to be metallic (Hulliger & Ott, 1977). Thus compounds with this structure fall into two classes with La₂Sb representing one class and Eu₄As₂O and Ba₄As₂O another; both Eu and Ba are divalent whereas La is trivalent.

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The structures of Eu₄As₃ and La₄Sb₃.* By Y. WANG,[†] L. D. CALVERT[‡] and J. B. TAYLOR, Chemistry Division, National Research Council of Canada, Ottawa, Canada K1A 0R9

(Received 4 September 1979; accepted 24 September 1979)

Abstract

Ln₄X₃ (Ln = rare earth, X = P,As,Sb,Bi) compounds have the anti-Th₃P₄ structure with space group $I\overline{4}3d$ (No. 220), Z = 4 and Ln atoms in 16(c) x,x,x and X atoms in 12(a) $\frac{3}{8}$,0, $\frac{1}{4}$. Eu₄As₃ has $a = 9 \cdot 214$ (1) Å, x (Eu) = 0.06798 (3), B(Eu) = 1.19 (1), B(As) = 1.33 (6) Å², $D_x = 7.07$ Mg m⁻³, Eu-(3 + 3)As, As-(4 + 4)Eu = 3.042 (1), 3.348 (1), Eu-3Eu = 3.581 (1) Å. La₄Sb₃ has a = 9.649 (1) Å, x(La) = 0.07063 (2), B(La) = 0.83 (1), B(As) = 0.81 (2) Å², $D_x =$ 6.81 Mg m⁻³, La-(3 + 3)Sb, Sb-(4 + 4)La = 3.211 (1), 3.476 (1), La-3La = 3.720 (1) Å. The Ln-X distances are normal. The Ln atoms form two distinct enantiomorphic three-dimensional networks composed of three-connected ten-membered rings. These Ln-3Ln distances correspond to the values expected for Ln³⁺ metallic bonds.

Introduction

Detailed structures of the lanthanon pnictides $(Ln_4 X_3: Ln = rare-earth element, X = Group V element)$ with the anti-Th₃P₄-type structure are of considerable interest in connection with their physical properties, *e.g.* ferromagnetism and superconductivity (Hulliger & Vogt, 1970; Hulliger & Ott, 1977). A study of the Eu-As system (Taylor, Calvert, Utsunomiya, Wang & Despault, 1978) found that the powder intensities were not sensitive to the As vacancy concentration. Accordingly, single-crystal structure studies

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were undertaken to define the composition more precisely. Crystals were extracted from the available specimens and tested by Laue photographs. After many tests, one reasonable crystal was found in a specimen of composition Eu₄As₃, melted and annealed in Ta at 1923 K and then rapidly cooled. Subsequently a second structure was determined for a La₄Sb₃ crystal from a specimen of composition La₄Sb₃ melted and annealed in Ta at 2173 K and then slow cooled. Intensity data were collected using the profile-analysis technique (Grant & Gabe, 1978) on a four-circle computer-controlled diffractometer with graphite-monochromated Mo $K\alpha$ radiation ($\lambda = 0.70932$ Å) with local programs (for details see Gabe & O'Byrne, 1970; Grant & Gabe, 1974;

Table 1. $\operatorname{Ln}_4 X_3(I\overline{4}3d)$

	Eu ₄ As ₃	La ₄ Sb ₃
a (Å)	9-214(1)	9.649(1)
V (Å 3)	782-4	898-3
x for Ln	0.06798 (3)	0.07063 (2)
U_{11}, U_{13} for Ln*	0.0151 (1), 0.0004 (1)	0.0105(1), -0.0004(1)
$U_{11}U_{13}$ for X^{\dagger}	0.0151 (4), 0.0204 (8)	0.0095 (1), 0.0119 (2
R,, R,‡	0.028, 0.021	0.021, 0.016
Total independent reflections	416	404
Observed $ I > 2\sigma(I) $	362	388
$\Sigma \langle \Delta I \rangle / \Sigma \langle I \rangle$ (%)	2	3
Extinction coefficient		8.97 × 10 °
$2\theta_{max}$ (°)	80	75
Scan range (2θ) (°)	$1.6 + 0.7 \tan\theta$	1·4 + 0·7 tanθ
Crystal size (mm)	$0.12 \times 0.04 \times 0.02$	$0.14 \times 0.07 \times 0.04$
μ (mm ⁻¹)	45.02	22.76
Transmission coefficient	0.26-0.39	0.18-0.33
F(000)	1404	1524
$D_x (Mg m^{-3})$	7.07	6.81

* For Ln, $U_{11} = U_{22} = U_{33}$ and $U_{13} = U_{23} = U_{12}$. † For X, $U_{11} = U_{22}$ and $U_{12} = U_{13} = U_{23} = 0$.

 $tR_1 = \Sigma |\Delta F| / \Sigma F_{\varphi}; R_2 = (\Sigma w |\Delta F|^2 / \Sigma w F_{\varphi}^2)^{1/2}; w = 1 / [\sigma^2(F_{\varphi})_{\text{counts}}].$

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Wang, Gabe, Calvert & Taylor, 1976; Larson & Gabe, 1978). The space group was determined from precession photographs (hkl: h + k + l = 2n; hhl: 2h + l = 4n) and intensity statistics to be I43d. The systematic absences were also measured to check the possibility that the crystal had lower symmetry due to ordering (Carter, 1972). In addition, the La_4Sb_3 data were refined in $I2_13$ to a result not significantly different from $I\bar{4}3d$. Details for each crystal are given in Table 1. Two complete unique sets, each containing *hkl* and $h\bar{k}\bar{l}$, were measured and averaged. All calculations were carried out on the laboratory computer using the NRC-PDP-8 system of programs (Larson & Gabe, 1978). The structure was refined using the observed reflections with allowance for anomalous dispersion and isotropic extinction (Larson, 1969). For Eu₄As₃ extinction was negligible and not refined. Refinement was by anisotropic full-matrix least squares with starting coordinates from Hulliger & Vogt (1970).*

For both structures the final difference maps contained no significant details. (A refinement with variable As occupancy did not give significantly different results.) A microprobe analysis (Owens, 1978) of the Eu-As crystal gave $Eu_{4,01(3)}As_{2,99(3)}$. The stoichiometric formula was therefore accepted for our crystal.

For comparison, La_4Sb_3 , known to be stoichiometric (Gambino, 1967), was also studied.

Discussion

The Th₃P₄ structure (Meisel, 1939) has been described in detail elsewhere (Heim & Bärnighausen, 1978, and references therein). The anti version of this structure is observed for pnictides of the rare earths, Ln_4X_3 (Hohnke & Parthé, 1966; Bodnar & Steinfink, 1966; Gambino, 1967; Hulliger & Ott, 1977), but to date no diffractometer singlecrystal study is known to us. Structural details are of interest in discussing the superconductivity of Ln_4X_3 compounds which are an order of magnitude different from the analogous $Ln_{3}Y_{4}$ (Y = S, Se, Te) compounds with the normal Th₂P₄ structure. The interatomic distances are given in the Abstract and the average Ln-X values, viz Eu-As = 3.20 Å and La-Sb = 3.34 Å, are equivalent to those expected for Ln^{3+} metallic phases. Each Ln atom is connected to three other Ln atoms (Maas, 1970), forming two distinct enantiomorphic sets of three-connected ten-membered rings (Heim & Bärnighausen, 1978) in which the bonding is comparable to that for the Ln³⁺ metal. For La these are the face-centred cubic and double hexagonal close-packed forms of the element (3.75 and 3.77 Å respectively) and for Eu³⁺ the 12coordination value is 3.60 Å. Similar geometrically distinguishable Ln sublattices are observed in other lanthanon pnictides, even anion-rich ones, *e.g.* the high- and lowtemperature forms of LaAs₂, and in NdAs₂ (Wang, Heyding, Gabe, Calvert & Taylor, 1978). The lattice parameter and stoichiometry found for La₄Sb₃ are in excellent agreement with those of Hulliger & Ott (1977), 9.6485 (8) Å and La₄Sb_{2.94-3.02}, and consistent with those of Hohnke & Parthé (1966), 9.648 Å, and Gambino (1967), 9.63 Å. For Eu₄As₃ both lattice parameter and stoichiometry differ from those given by Hulliger & Vogt (1970) and we conclude that our single crystal is stoichiometric Eu₄As₃.

We are very grateful to Mr D. R. Owens of the Mineralogy Section, Physical Sciences Laboratory, CAN-MET, for the microprobe analyses.

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On behalf of the International Union of Crystallography the

Executive Committee wishes to record its gratitude to the Rigaku Corporation, Tokyo, Japan, for a generous donation to the General Fund of the Union.

^{*} Lists of structure factors for both compounds have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34788 (10 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.